

# Potential Application of Modified *Saccharomyces Cerevisiae* for Removing Lead and Cadmium

### Lívia de CF\*, Mario HB and Benedito C

<sup>1</sup>Departamento of Microbiology, University of Sao Paulo, Sao Paulo, Brazil

#### Abstract

Recently, heavy metal pollution has become a worldwide concern. Saccharomyces cerevisiae have been used for bioremediation process for heavy metal uptake. In this study, *S. cerevisiae* was genetically modified with the gene EC20 (synthetic phytochelatin) with the purpose to enhance the accumulation capability for Lead ( $Pb^{2+}$ ) and Cadmium ( $Cd^{2+}$ ). *S. cerevisiae* has showed high biosorption capability of  $Pb^{2+}$ , followed by  $Cd^{2+}$ . Indeed, the transformed cells exhibited even higher capacity to accumulate lead. TEM (transmission electronic microscopy) images of transformed *S. cerevisiae* showed accumulation of heavy metals on cell wall, which was also verified by EDS (energy dispersive spectrum). There are many mechanisms involved in metal uptake and for some unknown reason. The results of yeast expressing EC20 were better detected when yeast cells were grown in the presence of  $Pb^{2+}$  than with  $Cd^{2+}$ , as clearly observed with the improvement in  $Pb^{2+}$  retention by yeast cell wall.

**Keywords:** Yeast; Genetic modification; Transformation; Heavy metals; Cadmium; Lead uptake

#### Introduction

In recent years, pollution of wastewater has increased and heavy metals (such as lead and cadmium) are one of the heavy metals in the environment, causing health effects [1-4] and serious environmental problems [5-8]. Mining, industrial and volcanic activities are the main source of heavy metals in environment [9]. In the aquatic system, heavy metal pollution has become an environmental concern, as it is extremely recalcitrant and non-degradable. Even at low concentrations, heavy metals can cause toxicity to humans and other forms of life [5]. It is necessary to uptake these heavy metals from wastewater before its disposal. The toxicity of heavy metals could be transformed from relevant low toxic species into more toxic forms. Moreover, most of the heavy metal salts are soluble in water, hindering some methods of separation (such as, chemical oxidation or reduction, electrochemical treatment, evaporative recovery, filtration, chemical precipitation, ion exchange, membrane technologies, adsorption on activated carbon) [10]. As a result, microbial biomass is an option for developing economic and eco-friendly wastewater treatment process [11,8]. An alternative process is called biosorption that can be defined as the removal of metal from solution by biological material (biosorbents) [12]. These biosorbents can effectively sequester dissolved metal ions of solutions with high efficiency and speed. It also offers several advantages over conventional treatment methods including cost effectiveness, minimization of chemical/biological sludge and regeneration of biosorbent with possibility of metal recovery [13]. Metal polluted environment contains fungi, which has adapted to toxic concentration of heavy metals and become metal resistant [14]. Metals retention by living fungi depends on several factors such as culture conditions and metal concentration [15]. Especially, fungi used for biosorption can be fulfilled by simple physical methods without any damaging the fungi structural integrity [16]. Saccharomyces cerevisiae can be easily cultivated using unsophisticated fermentation techniques and inexpensive growth media [17] at large scale or collected from waste of food industries and have become a benchmark in the study of biosorption. Moreover, it is an excellent model microorganism in biological research and its biomass is also considered as safe [18-20]. Therefore, biosorbents made from S. cerevisiae can be easily accepted by the public when applied especially to investigate the interactions of metal-microbe at molecular level [21]. The use of yeasts in model systems is particularly attractive because of the easy genetic manipulation and the availability of the complete genomic sequence knowledge on the molecular biology of the yeast is helpful to identify the molecular mechanism of biosorption in metal ion removal [22-24]. Transformation technique is an important tool in which exogenous DNA is introduced into a cell, resulting in genetic modification. To improve the biosorption capacity of the potential yeast biosorbent, a short metal-binding can be displayed in the cell's surface with purpose to increase its heavy metal biosorption capacity from solutions. Among the heavy metal-binding in the cells, the phytochelatins (PCs) is one of the best studied so far [25]. The EC20 gene encodes a synthetic phytochelatin [25-27]. These peptides bind to metal ions, keeping them on their cell surface [28,29]. To verify the capacity of binding heavy metal on the fungal cell, progress has been made in exploring the cell surface ultra-structure of yeasts owing to the development of techniques such as the transmission electron microscopy with energydispersive spectrum analysis (TEM-EDS) [18].

The aim of this study was to modify by genetic transformation a strain of *S. cerevisiae* to improve the biosorption of toxic heavy metals (Lead and Cadmium) and to compare the strain's capacity of retaining heavy metals before the genetic transformation and after that.

#### Materials and Methods

# Saccharomyces cerevisiae and Escherichia coli strains and growth media

S. cerevisiae W303-1A (MATa ade2-1 trp1-1 his3-1,15 leu2-

\*Corresponding author: Lívia de CF, Lívia de Carvalho Fontes, Departament of Microbiology - University of Sao Paulo, Brazil. Tel: +55-11-30917295; E-mail: liviacfontes@gmail.com

Received September 08, 2015; Accepted October 07, 2015; Published October 10, 2015

Citation: Lívia de CF, Mario HB, Benedito C (2015) Potential Application of Modified Saccharomyces Cerevisiae for Removing Lead and Cadmium. J Bioremed Biodeg 6: 313. doi:10.4172/2155-6199.1000313

**Copyright:** © 2015 Lívia de CF, et al. This is an open-a ccess article distributed under the terms of the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original author and source are credited.

3,112 ura3-1  $\rho$ +can<sup>R</sup>) strain (R. Rothstein-Columbia University) was cultivated on rich glucose media (YPD - 1% yeast extract, 2% peptone, 2% dextrose) and minimal glucose media (0,67% yeast nitrogen bases, 2% glucose) supplemented with the appropriate auxotrophic supplements. The *E. coli* strain RR1 ( $\Delta$ (*gpt-proA*)62, *leuB6, thi-1, lac*Y1, *hsd*<sub>B</sub>20, *rpsL*20 (*Str<sup>x</sup>*), *ara*-14, *galK*2, *xyl*-5, *mtl*-1, *supE*44, *mcrB*<sub>B</sub>) (3183) was cultivated on LB plate (Luria Bertani) supplemented or not with ampicilin for plasmid selection.

#### EC20 synthesis, vectors construction and yeast transformation

EC20 was assembled from the oligonucleotides pairs EC20-1 and EC20-2 [30-32]. The complementary oligonucleotides were heated at 90°C for 2 min and hybridized, first at 55°C for 30 min followed by 37°C for 30 min. The double stranded fragment was digested, respectively, with BamHI and HindIII and inserted into YIp351 [33] and YCp22 containing the *TRP1* marker [34] previously cut with the same restriction enzymes. YIp351-EC20 and YCp22-EC20 recombinant plasmids transformed W303-1A yeast strain following the procedure of Schiestl and Gietz [35]. YIp351-EC20 was linearized with BstXI and integrated into chromosomal *LEU2* locus [30].

#### **Recombinant DNA manipulation**

Standard methods were used for recombinant DNA manipulations, *E. coli* growth and transformation [36]. The recombinant plasmids YIp351-EC20 and YCp22-EC20 were isolated from *E. coli* in small and medium scale procedures [37,38]. For plasmid sequencing was used the Pure Yeild TM Plasmid Mini prep System kit (Promega Co., Madison, WI, USA). The obtained sequences were analyzed using Bio Edit program.

#### Metal ion solutions and biomass preparation

Two kinds of metal ion solution were used: solutions containing lead and cadmium. Metal ion solutions were made by dissolving analytical  $Pb(CH_3COO)_2$ .  $3H_20$  (1g/L) and,  $CdCl_2$ .  $2,5H_2O$  (0,5g/L) in deionized water. The fungal biomass was prepared in Sabouraud broth (Sb) (Oxoid, England), and incubated at 25°C for 5 days, and shaked at 150 rpm. After incubation, the pellets were harvested and washed with double-deionized water, corresponding to live biomass. The pH value of the solutions was adjusted. The conditions were optimized during the preliminary experiments to have all metals in a dissolved aqueous form and to achieve high biosorption capacity of all metal ions.

## Transmission electronic microscopy

From the growth of fungi, broth was filtered to obtain biomass that was sent to the microtome sector to obtain the ultrathin sections. The ultrathin sections were examined by electron microscopy (JEOL-1010) at IPEN and ICB department at the University of Sao Paulo. Transmission electron microscopy was performed with the strain of *S. cerevisiae* modified genetically and non-modified in this study with the purpose to reveal the presence of heavy metals in the fungal cell.

#### Results

# Transformation of Saccharomyces cerevisiae

With regard to transformation of yeast, *S. cerevisiae* strain was transformed with YCp22-EC20 and YIp351-EC20 recombinant. YCp22-EC20 was chosen for yeast plasmid rescue and it was again sequenced confirming the stability of YCp22-EC20 in yeast. The transformed cells were tested for increased ability to chelate heavy metals into their cell surface, which showed an improvement in

its capability to grow in medium amended with different lead and cadmium concentrations.

#### Transmission electronic microscopy

Transmission electron microscopy images and EDS analysis of the cells of S. cerevisiae showed that heavy metals uptake have increased by transformed yeast cells and also revealed the presence of heavy metals mainly in the cell wall. Comparing to the control (Figure 1), the cells showed a high increase in their Pb2+ binding capacity (Figure 2a and b). The micrograph Figure 2a showed a transformed S. cerevisiae after genetic transformation and the lead accumulated in the wall cell. The micrograph Figure 2b showed a non-transformed S. cerevisiae and the wall cell with less lead accumulated. As indicated above, we observe that S. cerevisiae cells have a lower capacity of Cd<sup>2+</sup> binding than Pb2+ binding (Figure 3a and 3b). TEM images and EDS analysis of S. cerevisiae grown in the presence of Cadmium after genetic transformation. The micrographs (Figure 2a) showed a transformed S. cerevisiae with cadmium accumulated in the wall cell. The micrograph (Figure 2b) showed a non-transformed S. cerevisiae and cell wall without cadmium accumulated. No heavy metals were detected in the control. As a result, during biosorption, the cells of S. cerevisiae released some substances without strongly damaging the cells integrity. The presence of electron-dense deposits was observed by transmission electron microscopy.

#### Discussion

Yeast cells have been used in the removal of heavy metal ions from wastewater [39-41]. *S.cerevisiae* could treat all major toxic heavy metals [1]. Metals are frequently internalized by the fungal cell and are accumulated into the vacuoles or can also be bound to low molecular weight proteins such as phytochelatins [41,42]. To improve metal uptake capacity, some techniques of biomass processing have been studied. Most biomass modification techniques described in the literature have shown to improve biosorption ability [43,44]. Metabolism-independent binding of metal ions in yeast cell is usually fast and large amounts of heavy metals can be bound to the cell [1]. In our study, *S. cerevisiae* cells were genetically transformed. Synthetic genes encoding for EC20 were synthesized as described previously [25,32]. According to our study, modified and non-modified strains were used to uptake lead and cadmium from broth. In studies of Bae [25], the EC20 gene was used to





Citation: Lívia de CF, Mario HB, Benedito C (2015) Potential Application of Modified Saccharomyces Cerevisiae for Removing Lead and Cadmium. J Bioremed Biodeg 6: 313. doi:10.4172/2155-6199.1000313





anchor EC20 on the surface of the cells, providing an increase in their heavy metal binding capacity. Furthermore, Biondo et al. [32] described the effects of the cell surface display of a synthetic phytochelatin in *Cupriavidus metallidurans* for enhanced metals bioremediation. Great progress has been made in exploring the cell surface ultra-structure of yeasts owing to the development of techniques such as the transmission electron microscopy with energy-dispersive spectrum analysis (TEM– EDS). In our experiments the majority of accumulated Pb<sup>2+</sup> was found in the cell wall, indicating the involvement of functional groups of cell wall [45-47], in sequestering the Pb<sup>2+</sup> ions. Moreover, Brady and Duncan [47] also found that biosorption can be achieved through complexation by cell wall and membrane hydroxylated components. Kunst and Roomans [48] found polyphosphate granules localized in or close to the cell vacuoles, which are the major storage vesicles for heavy metal ions in *S. cerevisiae*. Orlovich and Ashford [49] followed up on aforementioned studies and determined the exact vacuolar location for polyphosphate granules incorporation. Favero et al. [50] found that 80% of  $Cd^{2+}$  was bound to the hyphal cell walls in their study of *Pleurotus ostreatus*. In summary, the potential advantages in use of fungal biomass especially *S. cerevisiae* certainly could be applied for treatment of large scale heavy metals pollutants without the generation of toxic sludge. In the same context, genetic manipulation of *S. cerevisiae* could to increase the biosoption of heavy metals.

#### Acknowledgements

This work was supported by FAPESP (Fundação de Amparo à Pesquisa do Estado de São Paulo).

Citation: Lívia de CF, Mario HB, Benedito C (2015) Potential Application of Modified Saccharomyces Cerevisiae for Removing Lead and Cadmium. J Bioremed Biodeg 6: 313. doi:10.4172/2155-6199.1000313



Figure 3: TEM images and EDS analysis of *S. cerevisiae* grown in the presence of Cadmium. Cadmium in *Saccharomyces cerevisiae* after (A) and Before (B) Genetic transformation. The micrographs showed (2a) a transformed *S. cerevisiae*. On figure 2a, the arrow shows the cadmium accumulated in the wall cell. The micrograph (2b) showed a non-transformed *S. cerevisiae*. On figure 2b, the arrow shows the wall cell without cadmium accumulated. On both figures, EDS analysis confirms the presence/absence of cadmium on the local showed by the arrow.

#### References

- Thippeswamy B, Shivakumar CK, Krishnappa M (2014) Study on heavy metals biosorption ability of Saccharomyces cerevisiae. Int Journ Biol Research, 2: 106-115.
- Hu H, Jin Q, Kava P (2014) A Study of Heavy Metal Pollution in China: Current Status, Pollution-Control Policies and Countermeasures. Sustainability 6: 5820-5838.
- Martin S, Griswold W (2009) Human health effects of heavy metals. Environ Sci Technol Br, 15: 1–6.
- Ahalya N, Ramachandra TV, Kanamadi RD (2003) Biosorption of heavy metals. Res J Chem Environ 7: 71-78.
- 5. Monachese M, Burton JP, Reid G (2012) Bioremediation and Tolerance of

Humans to Heavy Metals through Microbial Processes: a Potential Role for Probiotics? Appl Environ Microbiol, 78: 186397-640.

- Bishop PL (2002) Pollution prevention: fundamentals and practice. Beijing: Tsinghua University Press.
- Wang JL (2002a) Immobilization techniques for biocatalysts and waterpollution control. Beijing: Science Press.
- Volesky B (1990a) Biosorption and biosorbents. In: Volesky B, editor. Biosorption of Heavy Metals. Florida: CRC press, 3-5.
- Sahni SK (2011) Hazardous metals and minerals pollution in India. Indian National Science Academy. New Delhi, 29.
- Tsekova K, Petrov G (2002) Removal of heavy metals from aquous solution using *Rhizopus delmar* mycelia in free and polyuretane-bound form. *Z. Naturforsch*, 57: 629-633.

- Veglio F, Beolchini F (1997) Removal of metals by biosorption: a review. Hydrometallurgy 44: 301-16.
- 12. Gadd GM (1993) Interactions of fungi with toxic metals. Phytologist, 124: 25-60.
- Shoaib A, Aslam N, Aslam N (2012) Myco and Phyto Remediation of Heavy Metals from Aqueous Solution. The Online Journal of Science and Technology 2: 3.
- 14. Ceci A, Kierans M, Hillier S, Persiani AM, Gadd GM (2015) Fungalbioweathering1 of mimetite and a general geomycological model for lead apatite mineral biotransformations. Appl Environ Microbiol.
- Melgar MJ, Alonso J, Garcia MA (2007) Removal of toxic metals from aqueous solutions by fungal biomass of *Agaricus Macrospores*. Science of the Total Environment 385: 12-19.
- Adeyemi AO (2009) Bioaccumulation of arsenic by fungi. American Journal of Environmental Science 05: 364-370.
- Kapoor A, Viraraghavan T (1995) Fungi biosorption-an alternative treatment option for heavy metal bearing wastewaters: a review. BioresourTechnol 53: 195-206.
- Chen C, Wenb D, Wang J (2014) Cellular surface characteristics of Saccharomyces cerevisiae before and after Ag(I) biosorption.
- Ferreira I, Pinho O, Vieira E, Tavarela JG (2010) Brewer's Saccharomyces yeast biomass: characteristics and potential applications. Trends Food Sci. Technol. 21: 77-84.
- Chen C, Wang JL (2010) Removal of heavy metal ions by waste biomass of Saccharomyces cerevisiae. J. Environ. Eng. 136: 95-102.
- 21. Zhou DQ (2002) Microbiology. (2ndedn). Beijing: Higher Education Press.
- 22. Eide DJ (1998) The molecular biology of metal ion transport in *Saccharomyces cerevisiae*. Annu Rev Nutr, 18: 441-69.
- 23. Eide D (1997) Molecular biology of iron and zinc uptake in eukaryotes. Curr. Opin. Cell Biol 9: 573-7.
- 24. Perego P, Howell SB (1997) Molecular mechanisms controlling sensitivity to toxic metal ions in yeast. Toxicol. Appl Pharmacol. 147: 312-8.
- 25. Bae W, Mulchandani A, Chen W (2002) Cell surface display of synthetic phytochelatins using ice nucleation protein for enhanced heavy metal bioaccumulation. Journal of Inorganic Biochemistry 88: 223-227.
- Bae W, Wu CW, Kostal J, Mulchandani A, Chen W (2003) Enhanced mercury biosorption by bacterial cells with surface-displayed MerR. Applied and Environmental Microbiology 69: 3176-80.
- 27. Bae W, Mehra RK, Mulchandani A, Chen W (2001) Genetic engineering of Escherichia coli for enhanced uptake and bioaccumulation of mercury. Applied and Environmental Microbiology 67: 5335-8.
- Cobbett C, Goldsbrough P (2002) Phytochelatins and metallothioneins: roles in heavy metal detoxification and homeostasis. Annu Rev Plant Biol 53: 159-182.
- 29. Mejare M, Bulow L (2001) Metal-binding proteins and peptides in bioremediation and phytoremediation of heavy metals. Trends Biotechnol 19: 67-73.

- 30. Rothstein RJ (1983) Methods Enzymol.101: 202-211.
- Hanahan D (1983) Studies on transformation of *Escherichia coli* with plasmids. Journal of molecular biology 166: 557-580.
- Biondo R, Silva FA, Vicente EJ, Sarkis JES, Schenberg ACG (2012) Synthetic Phytochelatin Surface Display in Cupriavidusmetallidurans CH34 for Enhanced Metals Bioremediation. Environ Sci Technol. 46: 8325-8332.
- 33. Hill JE, Myers A, Koerner TJ, Tzagoloff A (1986) Yeast. 2: 163-167.
- 34. Gietz RD, Sugino A (1988) Gene. 74: 527-534.
- 35. Schiestl RH, Gietz RD (1989) Curr. Genet 16: 339-346
- Sambrook J, Fritsch EF, Maniatis T (1989) Molecular Cloning: A laboratory manual, Cold Spring Harbor, NY: Cold Spring Harbor Laboratory.
- Birnboim HC, Doly J (1979) A rapid alkaline extraction procedure for screening recombinant plasmid DNA. Nucleic Acids Research, 7: 1513-1523.
- Ausubel F, Brent R, Kingston R, Moore D, Seidman J et al. (1989) Current protocols in molecular biology. [s.l: s.n.].
- Gadd GM (2009) Biosorption: critical review of scientific rationale, environmental importance and significance for pollution treatment. J Chem Technol Biotechnol. 84: 13-28.
- Wang JL, Chen C, (2009) Biosorbents for heavy metals removal and their future. Biotechnol. Adv. 27: 195-226.
- Volesky B, May-Philips HA (1995) Biosorption of heavy metals by Saccharomyces cerevisiae. Appl Microbiol Biotechnol. 42: 797-806.
- Krauter P, Martinelli R, Williams K, MartinsS (1996) Removal of Cr(VI) from ground water by Saccharomyces cerevisiae. Biodegradation, 7: 277-286.
- Shroff KA, Vaidya VK (2011) Effect of pre-treatments on biosorption of Ni (II) by dead biomass of *Mucorhiemalis*. Eng in Life Sci. 11: 588-597.
- Yan G, Viraraghavan T (2000) Effect of pre-treatment on the of heavy metals on Mucorrouxii. Water SA. 26: 119-123.
- Gadd GM (1993) Interactions of fungi with toxic metals. New Phytol. 124: 25-60.
- 46. Ashkenazy R, Gottlieb L, Yannai S (1997) Characterization of acetone-washed yeast biomass functional groups involved in lead biosorption. Biotechnol Bioeng. 55: 1-10.
- Brady D, Duncan JR (1994) Binding of heavy metals by the cell walls of Saccharomyces cerevisiae. Enzyme Microb. Technol. 16: 633-638.
- Kunst L, Roomans GM (1985) Intracellular localization of heavy metals in yeast by X-ray microanalysis. Scan Electron Micros. 1: 191-199.
- Orlovich DA, Ashford AE (1993) Polyphosphate granules are an artefact of specimen preparation in the ectomycorrhizal fungus *Pisolithus tinctorius*. Protoplasma 173: 91-102.
- Favero N, Costa P, Massimino ML (1991) In vitro uptake of cadmium by basidiomycetes *Pleurotusostreatus*. Biotechnol Lett. 13: 701-704.