

Bio-limiting nutrient circulation in the ocean's oxygen minimum zones

Grace A. Abarike^{*}, Sibo Wang, Zhiguang Song

Department of Chemistry and Science, Guangdong Ocean University, Zhanjiang, China

Abstract

The ocean's oxygen minimum zone (OMZ) is known to be biogeochemical complex in terms of nutrient mass-balance and major element cycles. The recent reports suggesting that the ocean's OMZ facilitate about 25-50% loss of nitrogen in the ocean predominantly by anammox coupled with the recent identification of the major contribution of DNRA to oceans NH4+ flux, further complicates our understanding of the oceans nitrogen circulation and budget. Also, studies have shown that OMZ promotes the occurrence of sulfide plumes. Moreover, the redox sensitivity of the ocean's OMZ(s) such as their active sulfate reduction and sulfide oxidation potential makes them important regions for the cycling of phosphorus. However, as have been predicted that future expansion and intensification of OMZ driven by climatic change, still the understanding of this biological system remains a challenge with regards to its role in the overall chemical budget of the ocean.

In this regard, we aim to provide a comprehensive summary of the overall circulation patterns of some essential but biologically limiting nutrients such as nitrogen, sulfur, and phosphorus in this OMZ of the ocean. This will benefit future researches on predicting global budgets of these nutrients and their cycles related to the marine environment as it will help bring out a good understanding of the inextricable link among each of these limiting nutrients, as have been researched and reported in varied ways.

Keywords: Bio-limiting nutrient, Nitrogen, Phosphorus, Sulfur, Oceans oxygen minimum zone(s)

Introduction

Dissolved oxygen concentration (DOC) is one of the most important indexes for the thriving of marine ecosystems and a key factor in determining the energy direction in the marine trophic levels. DOC is equally the power house for biogeochemical cycles, ocean nutrient circulation, biological production near the surface, and the degradation of organic matter in the oceans interior. However, certain layer especially the midlayers of the present-day ocean is characterized by a low level of dissolved oxygen [5]. DOC decline in the mid-layerof thewater column arises when organic matter oxidization exceeds its availability in less ventilated zones, thus from a hypoxic layer commonly referrd to as the oxygen minimum zone (OMZ)[5]. The emergence of the term OMZ since decades from Joel D. Cline and Richards,(1972), following early reports of a pelagic minimum oxygen concerntrations within the Pacific Ocean off Panama and to a more serious situation of no "oxygen at all" has attained immenrse recognition and importance in recent times. This immerse recognition of the OMZ is as a result ofits revealed expansion, intensity, and its role as a key to understanding the present unbalanced and altered nitrogen, carbon iron, phosphate cycle etc. The increase in hypoxic layers of the ocean is a direct consequence of global warming that causes the decline in oxygen solubility, concentration, and its vertical exchange in the ocean, thereby slowing down the oceans interior ventilation through propagated stratification. As nitrate is depleted or used up, sulfate dominated reducing microbes assume, controlling the biogeochemical process involved in the cycling of marine sulfur. Also, hypoxia facilitates the greater recycling of phosphorous (P), from sediments.

The oxygen minimum zone of the ocean occurring in mid-layer of the water column is hostile and inhabitable to metazoan life but is characterized by declined high trophic level diversity but abundant pelagic microfauna. The microbial community in the OMZs possess assemblages of diverse genetic repertoire of putative sulfate, nitrate, and phosphate reducing bacteria whose combined metabolism plays intimate roles in its biogeochemical cycles and therefore which permits and enhances the use of alternative electron donors and acceptors during energy metabolism. Specifically, OMZs are harbored by bacteria and archaea that arbitrate and facilitate the loss of oceanic fixed nitrogen to the atmosphere through denitrification and the recently discovered anaerobic oxidation of ammonia to N_2 (anammox).

This review presents an overview of the cycling of some basic vital marine elements (nutrients) and how the overall declines in ocean oxygen concentrations affect their circulation, balance, and availability in the ocean based on published reports. It's hoped that the summarized information in this article would help improve our understanding of OMZ involvement in the circulation of these nutrients. This may facilitate a realistic assessment of the short and long-term impacts of ocean de-oxygenation on the cycling of these nutrients in the ocean's OMZs, as well as predict the future budgets of these nutrients in the ocean.

Bio-limiting nutrient circulation in the ocean's OMZ

The availability of bio-limiting nutrients such as nitrogen, phosphorus, and sulfur other than H_2O and carbon, are necessary supporting nutrients required for primary and biological production. Biological limiting nutrients can be absorbed by primary producers via roots from the soil and in aquatic systems directly from solution/water to boast primary production. However, the availability of these essential nutrients in the ocean

*Corresponding author: Grace AA, Department of Chemistry and Science, Guangdong Ocean University, Zhanjiang, China, E-mail: star_song@yahoo.com

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is threatened by the decline in DOC which has been observed to be facilitated by temperature increase. In addition, observations from four modeled based assessment, it was shown that global warming caused declines in marine productivity at lower altitudes, suggesting that the recycling of these nutrients in marine environments is undoubtble affected by stratification of the water column. Reported that nitrogen limitation affected primary production in eastern tropical north atlantic (ETNA) and eastern tropical south pacific (ETSP).Despite the numerous reports on the effects of global warming on nutrient circulation and marine productivity, there is still low confidence and insufficient measurement of the present, and predictions of the future net primary productivity of the oceans. Also, there is no clear distinction of climate induced signals on primary production and that of those resulting from natural variability as have been explained earlier.

Nitrogen circulation in the ocean's OMZ

Nitrogen plays a major role in the biogeochemistry of the marine environment, with considerable influence on the cycling of other elements, particularly carbon [4] and sulfur oxygen concentration which is determined by the O_2 supply and remineralization of sinking particulate organic carbon (POC). IN oxygen sufficient regions of the ocean, nitrification, a biologically enhanced process converts NH₄+ to NO₂- and NO₃-, whereas in anoxic ocean regions, nitrate, is generally known to be reduced in a stepwise manner to N₂O or N₂ (2NO₂ \rightarrow 2NO₂ \rightarrow 2NO \rightarrow N₂O \rightarrow N₂) for the respiratory oxidation of organic matter. Though, emphasized to be a hostile environment for almost all marine life[15], and cover about only about 7% of the oceans, global estimates show that the oceans oxygen minimum zones (OMZs) , promote about 25-50% of the removal through anaerobic microbial activities especially anammox denitrification, of oceanic fixed nitrogen. Thus, OMZs, are known to be the focal zones of oceanicnitrogen loss through denitrification and may cause lower nitrate concentrations in upwelled water (Löscher et al., 2016). Studies of the past geological records of sediments provides evidence of the important roles OMZ [4] and coastal-offshore OMZwater mass exchange play on the overall N budget. The role OMZs play in the nitrogen circulation especially its loss through heterotrophic and anammox denitrification has been evinced through several in-situ experiments. The diversified decadal measurement of the rate of nitrogen loss in the OMZof oceans, employing methods, such as direct quantification of nitrogen gas, its ratios with Argon (N,:Ar), stoichiometric approaches, and electron transport activity assays etc. For areas such as those of the ETSP [15], ETNP, and the Arabian Sea are employed. However, none of these methods has been able to distinguish N₂ production resulting from anammox denitrification and that of heterotrophic denitrification. Also the rate and regulatory pattern involved in the production and release of N_2 by each of these processes yet to be established.

Anammox denitrification is also known as bacterial oxidation of ammonium with nitrite under anoxic conditions (anammox) to produce N₂ (NH₄+ + NO₂- \rightarrow N₂+ 2H₂O). During N-cycling processes, the substrates required for anammox denitrification, NH₄+ and NO₂- can be obtained from reduction of NO₂ through heterotrophic denitrification processes and other possible sources such as nitrification (nitrate reduction to nitrite or the oxidation of NH₃) and the oxidation of organic matter through dissimilatory nitrate reduction to ammonium (DNRA) ,(NO₃ - / NO₂- to NH+4) and possibly excretion by zooplankton though remains to be established. Observations from the Peruvian and the Arabian Sea OMZs shows that NO₃ - reduction significantly correlated with depth-integrated NO₂- concentrations which suggests that that NO₃. reduction is a major contributor to the secondary NO₂-. Similarly, an annual estimation of about 51% and 65% of NO₂- oxidation from coastal and offshore

OMZ, respectively, of NO₂- reduction of the ETSP was discovered This observed occurrences, in the OMZs, suggests active N-loss in the ocean with the DNRA pathway, the anammox bacteria through multi haemnitrite reductase convert nitrates into nitrites and eventually to ammonium which is then combined with nitrite to form nitrogen gas. In early studies of the relative contributions of these process to marine nitrogen losses estimated through sealed sediments /water amended with 15N-labelled nitrogen species, DNRA was assumed to be insignificant compared with heterotrophic and anammox denitrification as anammox bacteria can also produce 15N15N from15N nitrate, even in the absence of dissimulator nitrate reducers. The combined reactions of these reactants with anammox could be derived from the recovery of the label in the produced N₂ and may result in the production of a double 15N labeled N₂ (15N15N), which in recent studies is tagged to complicating the analysis of the data during isotope-pairing experiments. However, [4]identified in the anoxic Peruvian shelf and upper slope sediments that, DNRA contributed upto 80 % to the total benthic NH,+ release as the major proportion of the total NO₃- + NO₂- uptake was routed into DNRA. Likewise, Sommer et al., (2016) estimated that DNRA at the shallowest station of the Peruvian contributed upabout63 % (12.4 mmol m-2 d-1) to the total NH₄+ flux. In OMZ off the coasts of Peru and Namibia, DNRA has been found to occur in regions where denitrification was not detectable but low rates of DNRA was detected, but also undetected in ETSP. However, the potential of DNRA to occur in open waters is still insufficiently explored and the distinction between the supply of NH₄+ and NO₂- from DNRA and the other mentioned sources have not been extensively explored.

Aside the contributions of DNRA, nitrification etc. processes to the thrive of anammox denitrification in oceans OMZ, ex-situ sulfide incubation experiments within the Chilean OMZs suggest that cryptic sulfur cycling contributes about 30% of the of carbon mineralization and the subsequent release of NH4+ for anammox (Johnston et al., 2014). There has also been a recent discovery of (OMZs) as a potential niche for NO₂dependent n-damo, a methanotrophic and sulfate-dependent bacteria responsible for anaerobic oxidation of methane whose pathway reduces NO_2 - to nitric oxide (NO), which is then putatively dissimulated into N_2 . These imaging scenarios coupled with the predicted future expansions and intensification of the oceans OMZs; suggest that the future global losses of nitrogen may increase. However, a confirmation of the potential for the activities of the n-damo in the pelagic OMZ and the inclusion of these nitrogen loss contributing factors/ activities in assessment of global nitrogen budgets is still lacking. Also, despite the significant effects nitrogen losses imprints in seawater nutrient, especially ratios of nitrogen to phosphorus and reduces the availability of nitrates for primary producers such as phytoplankton. However, these studies are still speculative, as biogeochemical models do not reproduce present-day global patterns of N. Also, with the existence foam in the oceans and its predicted vertical expansions in the near future in response to global warming we suggest the possibility of a growing and thriving mass of the denitrifying anammox bacteria, therefore we recommend further studies in other to determine the relative global distribution of this bacteria which is seldom studied.

Sulfur circulations in the ocean's OMZ

Regions of the oceans that are characterized by extreme hypoxia or worse cases of anoxia such as the OMZ of the ocean, facilitates the degradation of organic carbon leading to the production of hydrogen sulfide (H₂S), from the heterotrophic mineralization of organic matter by giant sulfate (SO₄₂-) reducing bacteria. Under well balanced circumstances, such as a steady state where the diffusive fluxes of nitrate and sulfide

are in a 1:2.5 ratio with sufficient concentrations O₂, sulfide-oxidizing, and nitrate-reducing these bacteria are capable of detoxifying sulfide into sulfur, preventing its release into the water column and reducing the frequent occurrence of sulfidic events. On the other hand, in the oceans oxygen minimum zones, observed sulfide, a product of reduced sulfate often originates from sediments of nitrate and nitrite depleted waters that act as terminal electron acceptors for the oxidation of H₂S as the reduction of NO₃- and NO₂- to NH₄+ is accompanied by oxidation of dissolved sulfide (H₂S) to sulfate (SO₄₂-) via elemental Sulfur as a reactive intermediate. Under these nitrate and oxygen deficient conditions, the accumulated concentration of sulfide reaches higher concentrations as it diffuses into the pelagic water column. Demonstrated in a time series station in the Baltic Sea, a possible occurrence of frequent sulfidic plumes events which is likely to also occur in other eutrophication and OMZs hot zones such as the Atlantic and Pacific oceans. The massive Beggiatoa and Thioploca microbial communities harbored by the OMZ of the ocean e.g. OMZ sediments of the Arabian Sea, the Benguela current ecosystem off Namibia and that of the Basaltic sea etc. significantly reduce the sulfide flux from the sediments to the overlying water column. The quantified release of this toxic substance (H₂S) especially from the underlying sediments into the water column can lead to the occasional build-up of high concentrations of H₂S in bottom water [7] e.g. diatomaceous sediments of Benguela current upwelling system and the SO42- reduction by pelagic microorganisms within the water column enhances its buildup in oceanic waters e.g. Black sea [1], promoting the loss in marine life and deterioration of the marine ecosystem at large. Also under overextended conditions of declined bottom water oxygen (O₂), NO₃- and NO₂-), microbial sulfide production can overcome the sulfide oxidation capacity of the sulfur bacteria and lead to marine sulfide poisoning of life. In an attempt to determine the nitrate and nitrite concentration in the Peruvian OMZ discovered that, sediments had a high potential for sulfide release and began to release sulfide as soon as NO₃- and NO₂- become depleted. Furthermore the results from pore water geochemical and non-steady state model techniques employed by to determine how sediments with nitrate-storing bacterial communities respond to NO₃- and NO₂- depletion, predicted an almost immediate increase in sulfide fluxes when NO₃and NO₂- where exhausted.

The metabolism of Sulphur is considered primitive and genes related to its oxidation and reduction is dispersed throughout the bacterial and archaeal domains [12]. Sulfur-oxidizing bacteria may be linked to that of the dissimilatory metabolism of oxidized nitrogen species as genomic analysis of the lineage SUP05, a free-living gammaproteobacteria relative of clam endosymbiosis from the a North Pacific seasonal OMZ, revealed enzymes necessary for the chemolithotrophic oxidation of reduced sulfur, as well as those for nitrate reduction to nitrous oxide (N₂O)[9]. Furthermore, the discovery of the bacteria, Epsilonproteo in the deep sea are thought of as being involved in the cycling of marine sulfur. Advanced studies done, has shown that the bacteria Gamma may also be involved in the oxidizing of sulfur in the hypoxic marine water columns and both bacteria have an overlapping but slightly different distribution, potentially related to different redox potentials. Gammaproteobacteria were identified in the free nitrate-rich OMZ waters off the Chilean coast suggesting its involvement in the cycling of sulfur [10]. Likewise, in sulfidefree but nitrate-rich regions of the Saanich Inlet, sulfide-oxidizing and nitrate-reducing genes were found. Past studies have shown that the sulfur producing species have been reduced as they are predated and oxidized by heterotrophic, chemolithoautotrophic bacteria, reducing their threat to releasing hydrogen sulfide. However, a series of shipboard incubation experiments from current studies of waters obtained from OMZs demonstrates active sulfate reduction and sulfide oxidation. OMZ waters

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promotes pronounced sulfate reduction and likely responsible for substantial NH+4 production. However, concise evidence connecting sulfate reduction and NH+4 productions is still insufficiently explored. They also discovered (oxygen, nitrate, nitrite, nitric oxide, and nitrous oxide) utilizing and distinct chemolithoautotrophs, $\gamma\text{-},\,\delta\text{-}$ and $\epsilon\text{-proteobacterial taxa}$ involved in the oxidation or reduction of sulfate in what they expressed as the largest plume ever recorded in oceanic waters, that occurred at the OMZ off Peru. This could be a consequence of the climate change accelerating temperatures and ocean stratification, and subsequently facilitating the (H₂S)production rate, however direct field observations are lacking. Likewise, the activities of the Thioglobus . Perditus together with free-living SUPO5 bacteria in the Peru upwelling and their dispersal facilitated by mesoscale eddy-driven cross-shelf, was identified [9]. However, the cell abundances, distribution, metabolic activities/capabilities of this free-living SUPO5 bacterium in the global ocean is still insufficiently explored.

The sulfur cycle has been proposed as a partial solution to the biogeochemical problem of closing marine fixed-nitrogen budgets in OMZs. However, the recent discovery of the cryptic sulfur cycling which refers to the simultaneous activity of sulfate-reducing and sulfide-oxidizing pathways in a closely defined space such as a marine particle aggregate, has been attributed to be contributing to the loss of fixed N from productive upwelling regions, the production of climate-relevant N₂O, and dark carbon fixation in the sub-euphotic water column [9]. These reductions of the sulfate is said to onlybe active n the presence of thermodynamically fully utilized electron acceptors such as nitrate and nitrite, but [10] challenged this idea in their recentexperiments suggesting that an active, but cryptic sulfur cycle is present in non-sulfidic subsurface waters in the eastern tropical South Pacific OMZ off northern Chile and concluded from outcome of their calculations, that sulfate reduction is still a thermodynamically favorable process in these OMZ waters. However, more studies on this should be done to rule out these speculations. Also, speculations of other OMZs of the global oceans, to be reservoirs of actively coupled sulfur and nitrogen cycles [10] most especially the cryptic sulfur cycle must be addressed foraccurate global assessment of the balance of this nutrient in the ocean.

Phosphorus circulation in the ocean's OMZ

Phosphogenesis is responsible for the removal of terrestrial phosphorus in the form of phosphate by intense weathering of sedimentary basins accumulated with phosphorous. Phosphate is one of the biogeochemical limiting nutrients whose existing percentage determines how much carbon [3] is incorporated into the living biomass and how much is consumed during the growth of phytoplankton in the aquatic environment. In the marine environment, phosphorus rapidly undergoes recycling in the water column but some organic components of it, stretches to the sediments especially under hypoxic conditions such as OMZ, as particulate organic matter such as phytodetritus, including those absorbed by iron oxyhydroxides and are later degraded at a certain depth, releasing HPO₄₂, limiting their sedimentary deposition. The sedimentary phases of phosphorus (P) include but not limited to exchangeable P, ferric Fe associated P, authigenic, biogenic and CaCO₃-bound P, detrital P and organic P. Under extended burial conditions, the labile P converted to HPO_{42}^{-} and released from the sediments undergoes diffusions or precipitation as authigenic calcium phosphate minerals such as $Ca_{s}(PO_{4})3 - x(CO_{3})x(OH,$ F). These chemical forms of P in deposited sediments are a reflection of the redox and can be used to reconstruct the depositional conditions from sedimentary records. However, the burial efficiency of is altered by the concentration of benthic oxygen, as it influences the reduction of the iron oxyhydroxides in reducing surface sediments, which decreases the

capacity of the sediment to retain upward diffusing dissolved phosphate. It also alters the release from organic matter enhanced relative to carbon under reducing conditions.

Phosphates can be deposited as sedimentary rocks, formed from a series of biological actions coupled with mineral precipitations. Oceanic anoxic events [6] play a major responsibility in the substantial deposition of phosphorite from the geological past, usually from preferentially regenerated from P-bearing organic matter as compared to Carbon, such as black shale[15]. Much of the regeneration takes place in the water column but in sediments of relatively shallow water bodies. The decline in oceanic oxygen has a marked effect on the cycling of phosphorus as under hypoxic conditions, phosphate (TPO₄) is dissolved and released into the pore fluids and subsequently into the overlying bottom water. Iron oxyhydroxides absorb this released phosphate and reduce them through microbial induced dissolution and the associated liberation of metal-oxide-bound phosphate, into absorbable states (soluble states), all enhanced under oxygen-deficient conditions. Hydrogen sulfide may then further enhance the phosphate released from the Oxyhydroxides through reductive dissolution. Studies on the mechanisms of phosphorus cycling especially in the hypoxic region of the ocean are still insufficient, therefore, more laboratory, field predictive future phosphorus flux models and assessment of the distribution of phosphorus using eg. remote sensing etc. shloud be done to build a comprehensive understanding of this component of our environment. Phosphogenesis is likely driven by large sulfur bacteria from the filamentous, highly motile [6] genera Beggiatoa, Thioploca and non-motile Thiomargarita because of their marked frequent and large existence as fossils in phosphorite rocks [2] in active regions of modern phosphorite formations[6]. Likewise, the occurrence of phosphorus is also linked to the result of oversaturation in the pore water with respect to apatite induced by the activity of microorganisms such as large sulfur bacteria, Thiomargarita, and the phylogenetic ally and metabolically closely related genus Beggiatoa, that results in its precipitation [6].Dating back to 1986, [17] reported the release of phosphorus upon the addition of sulfide which acted as a stimulant. In additions, [6] found out from their quest to ascertain whether "acetate or sulfide is responsible for the release of phosphate by a marine Beggiatoa strain through the decomposition of polyphosphates" that phosphorite is formed preferentially in sediments with high sulfate reduction rates. Despite the discovery of the different sources and formation processes of phosphorite, a clear distinction of biologically available phosphorus and that of dissolved organic phosphorus from lateral fluxes and their significant potential transformation mechanisms, is still under discovered in hypoxic waters of the global ocean.

Regions of local upwelling, where sulfate reduction rates are exceptionally high are characterized by pronounced phosphogenesis [10] as they are formed in shelf sediments that are located at the border of oxygen-depleted water masses [6]. In occasions where bottom waters are oxygenated and nitrified, polyphosphate commonly referred to as poly-P and nitrates accumulated and stored in vacuoles, are speculated to be taken up during short periods. Consequently, when oxygen concentration drops to zero (anoxic), energy from the hydrolysis of the polyphosphate is gained by the W. Anoxic bottom-water conditions coupled with regions where sedimentary iron cycling plays a major role in organic matter mineralization and also promotes the benthic regeneration and reflux of inorganic phosphate to the water column [11]. Active phosphogenesis is mainly found beneath the nutrient-rich upwelling areas off the coasts of Peru, Chile and Namibia, in modern oceans, representing the major hotspots. However, during studies of microbial marine phosphorus cycle, the direct measurement of microbial phosphorus is difficult to achieve as for example, DNA sequences need to be measured directly or removed or its influence terminated during the, DNA sequence analyses of diagnostic P-cycle but this practices is lacking. Also, the global assessment of the gains and the losses of phosphorus together with their distinctive loss pathways are still rarely if there is any, explored in the marine oxygen-starved regions.

Conclusion and recommendation

Climate-induced ocean oxygen declines and the predicted future further declines threaten the balanced distribution of the ocean biodiversity as it controls and alters the cycling patterns of basic biologically limiting nutrients such as sulfur, phosphorus, and nitrogen. The sensitivity of N loss processes in response to minimal changes in O_2 potential for additional processes, such as preferential of sulfide and phosphate released directly in the water column drives the oceanic nutrient balance. The excess or limited production of these nutrients has a marked effect on the survival of the inhabitants of the marine environment and OMZ plays a major role in this.

However, global estimations of how much and at what rate sulfide is generated in the water column during a sulfidic event are not well resolved. Further investigation should be made to determine the distribution of these nutrient cycle altering organisms with depth and their specific niches. Also, the discovered combined effect of the eddy-driven cross-shelf transport and of UT. *perditus* to denitrify and thrive on elemental sulfur in the absence of dissolved sulfide should be further investigated. Much attention should also be given to the uprising cryptic sulfur cycle, to ascertain if there are any associations with the cycling of the other limiting nutrients such as carbon, phosphorus, etc. since it is linked with nitrogen cycling that in turn, has the influence of other biochemical cycles. Furthermore, global assessment of the gains and the losses of phosphorus together with their distinctive loss pathways are still rarely if there is any, explored in the marine oxygen-starved regions (OMZ).

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