



## Integrated Processing of Rice Hulls for Production of Polyfunctional Materials

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### Abstract

The following method has been proposed for processing the rice hulls with producing materials of polyfunctional purpose that includes the washing, drying, and pyrolysis of the raw materials, and condensation of the vapor-gas mixture. The solid product – nanocomposite produced by densely packed carbon nanoparticles (~500 Å) and silicon dioxide (100÷200Å), which were present in the amorphous form in number of ~52.0% and ~35.0% respectively, proved to be advanced filler for elastomers, sorbent for recovery of precious and rare metals, and fodder supplement for the farm poultry. The organic product (aqueous solution of carboxylic acids (22%), phenols (14%), ketones (12%), cyclic aliphatic hydrocarbons (4.5%), heterocyclic compounds (4%), and spirits and esters (4.5%)) is a high-selective collector to the lead minerals when beneficiating complex rebellious ores, plant growth stimulant used on the non-productive soils and an antiseptic agent. The mixture of non-condensable gases (percent by volume: CH<sub>4</sub> ~ 40÷45; C<sub>2</sub>H<sub>4</sub> ~ 4÷6; CO<sub>2</sub> ~ 20÷22; CO ~ 22÷24; H<sub>2</sub> ~ 4÷6) can be used for producing carbon black or as a high-calorific fuel

**Keywords:** Rice hulls; Nanocomposite; Carbon nanoparticles; Silicon dioxide; Filler for elastomers; Sorbent

### Introduction

The recovery of the rice hulls (RH) is a topical and complex problem, which the scientists of many countries of the world are working at over the period of many years. The list of the proven trends in the RH application is so great that it would seem that this problem should not exist in principle. But on the strength of a number of factors neither of the proposed technologies had the commercial implementation and acuteness of the assigned task continues aggravating.

The method of the thermal processing is the most prevalent method for the problem solving. Some alternative versions of the rice hulls thermal destruction have been known [1-3], which, unfortunately, have a number of disadvantages. The major of them is lack of utilization of the non-condensable gases, which are discharged into the atmosphere polluting the environment or are used as a fuel that precludes a potentiality to obtain valuable chemical by-products.

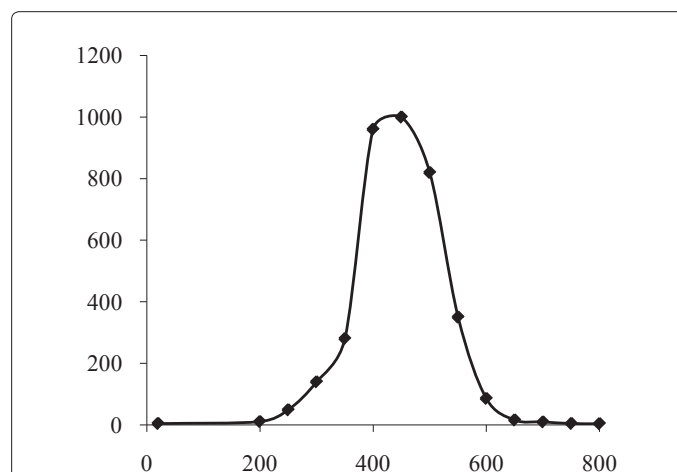
### Details of Study

The objective of this study was determined as follows: to develop the method for the rice hulls processing, which is free of the above said disadvantages, and to determine potential fields for the obtained products application. The tests object was the rice hulls of the Republic of Kazakhstan.

The rice hulls consisting of cellulose (33%), hemicellulose (18%), lignin (26%), moisture (1%), resinous matter (2%), neutrals (5%) and mineral substances (SiO<sub>2</sub> ~ 14%, the sum of CaO, MgO, K<sub>2</sub>O, Na<sub>2</sub>O, Fe<sub>2</sub>O<sub>3</sub>, Al<sub>2</sub>O<sub>3</sub> and others - 1%) were subject to thermal destruction at 650-750°C without free access of air in the rotary kiln that provided the uniform heating for the rice hulls. Pyrolysis of the material under review resulted in formation of the solid residue and pyrolysis gases, which represented a mixture of condensable vapor and non-condensable gases. The solid product was continuously discharged into the receiving bin. The formed vapor-gas mixture was forwarded to the condensation system. The condensable substances were accumulated in the special collector [4].

By the thermal analysis findings, it has been specified that the

maximum rate of the RH decomposition occurs in the temperature range 230-270°C. At the temperature of 300°C the formation of some advanced materials is developing, which is accompanied by high-heat evolution. The reactions of diminution and elimination of the loosely bound functional groups, as evidenced by rise in concentration of free-radical states (N) in the developed carbonizates, at 450°C are alternating with reactions of condensation of the solid-phase aromatic rings with formation of polycycles, i.e. graphite-alike grids, since N exactly at this



**Figure 1:** Changes in Concentration of Free Radicals in Pyrolysis Products With Temperature Rise.

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temperature, reaching the peak, then decreases (Figure 1), and in the respective IR-spectra the valence-vibrations band is indicated of the conjugated C=C-bonds in the range of 1600 cm<sup>-1</sup> (Figure 2).

The nanostructure of the carbon phase (G) was presented by crystallites with the cross sectional dimension  $L_{002} \sim 20-30 \text{ \AA}$ , which were composed as analogous graphite grids with the spacing C-C 1.42 Å. But as compared with the graphite structure, this nanostructure is less perfect [5].

By X-ray phase analysis method, in the composition of the rice hulls carbonizates besides the graphite phase (G) the presence of two hydrocarbon phases (Table 1) was detected. One of them is the polynaphthenic phase (N), which is diagnosable by the main reflection with  $d \sim 4.7 \text{ \AA}$ , represents the clathrate structure that comprises the condensed naphthene cycles separated by methylene bridges, which contain the alkane chains. The structure of the second hydrocarbon phase (H) with  $d \sim 8 \text{ \AA}$  is not identified in default of this phase in clear form (Table 1).

The rice hulls pyrolysis conditions exert also influence on X-ray characteristics of G in the carbon residues (Table 2): with the temperature rise the interlayer space  $d_{002}$  decreases, the average thickness of the formation members (height of crystallites)  $L_{002}$  and graphitization level  $c$  are increasing (it is not practical to calculate the

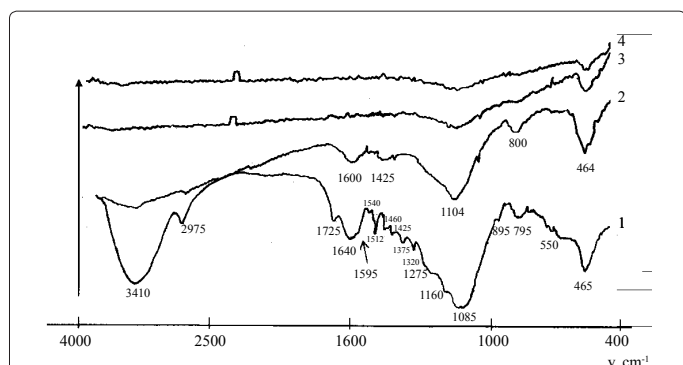


Figure 2: IR-spectra of the Original and Pyrolyzed Rice Hulls: 1 – Original RH; 2, 3, 4 – RH pyrolyzed at 450, 600, 800°C respectively

Pyrolysis temperature, °C	Phase content, %		
	G	N	H
500	45	45	10
650	55	35	10
800	60	30	10
1000	100	N/A	N/A

Table 1: Phase Composition of the Carbon-Containing Component in the Products from the Rice Hulls Processing.

Pyrolysis temperature, °C	Parameters for G		
	$d_{002}, \text{ \AA}$	$L_{002}, \text{ \AA}$	$c = \frac{L_{002} \times 10^{-3}}{d_{002} - 3.35}$
	$d_{002}, \text{ \AA}$	$L_{002}, \text{ \AA}$	
500	3.80	20	0.45
650	3.80	23	0.50
800	3.75	26	0.65
1000	3.75	30	0.75

Table 2: The X-ray Indices of the Graphite-like Phase of Carbonizates from the Rice Hulls.

Name of characteristic	SC
Iodine value, g/kg	54-58
Absorption of dibutyl phthalate, cm <sup>3</sup> /100	100-110
pH of the aqueous slurry	7-9
Mass fraction of losses at 105°C, %	2
Bulk density of the granular carbon, kg/m <sup>3</sup>	418

Table 3: The Physical/Chemical Characteristics of the Silicon-Carbon Material.

mean diameter of the bedded flat fragments of molecules  $L_a$  due to low intensity of reflections in the area of 21-22 °θ). These parameters behavior is indicative of the fact that the structural transformations of the materials under review are mainly developing for account of the intrablock and interlayer orientation.

By the Raster Electron Microscopy data (Figure 3), the rice hulls maintain their topology even at relatively high temperature action (up to 800°C) most probably due to high content of silicon, which as seen from the cluster of the light points in Figure 3c is mainly found in the outer and inner tissues of epidermis and also localized in definite places of the inner cells. However in this case the chemical elements redistribution takes place. As a result of the thermal destruction of the rice hulls organic component, a peculiar carbon skeleton is formed (Figure 3e), which is uniformly filled with the silicon-containing phase (Figure 3f) [6].

The Transmission Electron Microscopy results indicated that carbonizates of the rice hulls were formed by particles with different morphology among which the laminar type and translucency appear to be the common features (Figure 4).

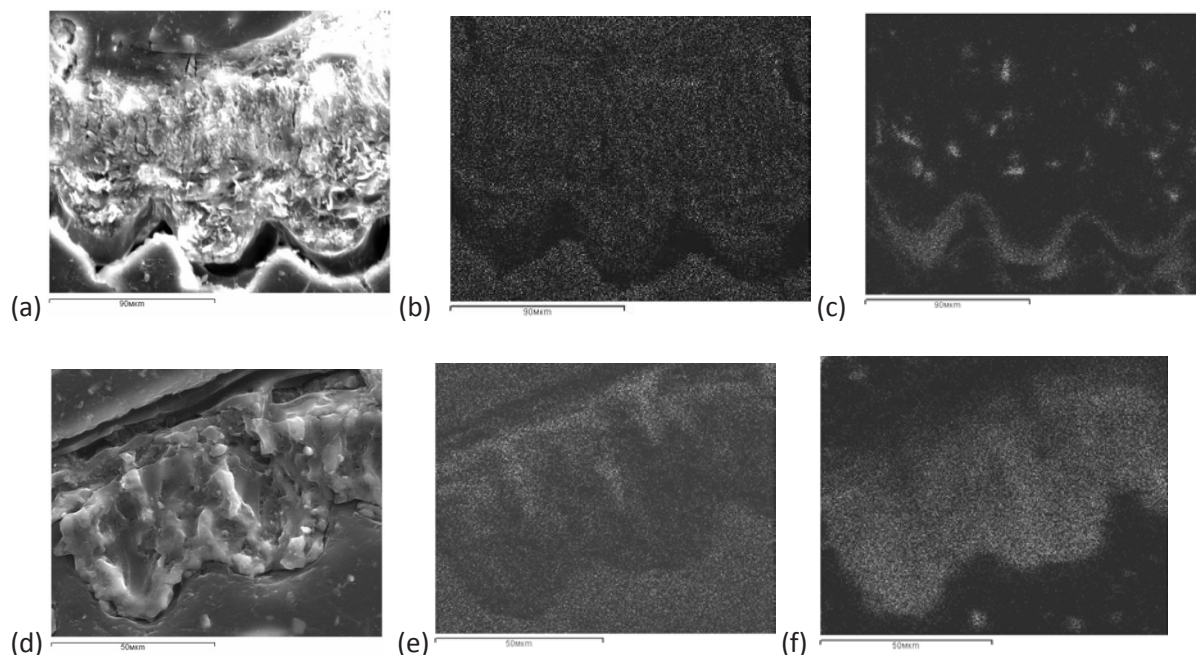
In virtue of the fact that SiO<sub>2</sub> in the rice hulls is in close connection with the organic components, in consequence of the plant tissue thermal destruction it forms a non-separate structure, and on account of high interparticle interaction with carbon it forms a nanocomposite assembly. Therefore, one may draw a conclusion that the solid product from thermal processing of the rice hulls represents nothing else than a nanocomposite, i.e. a macroscopic object formed by densely packed nanoparticles of carbon and silicon dioxide, which are present practically in equal quantities (50-55 and 40-45% by mass respectively) and uniformly distributed between them.

Due to the fact that the technical carbon is not produced in Kazakhstan, while it is the second ingredient after the natural rubber that defines the service performance of the rubber articles, and organization of such production facility seems unpractical considering necessity in combustion of expensive and exhaustible hydrocarbon materials (oil, gas and coal), in the process of which some carcinogenic emissions are originated, it was of some interest to evaluate the silicon-carbon material as a filler for elastomers.

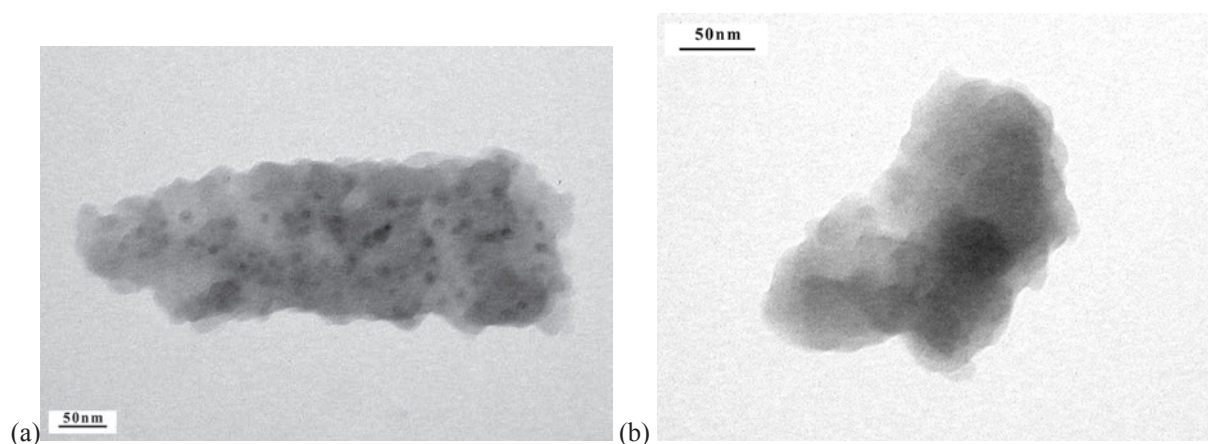
The physical/chemical characteristics of the silicon-carbon (SC) are given in Table 3.

The silicon-carbon product of two types: the product that contained the form of the original rice hulls with the grain size up to 5 mm (granular product) and the product crushed in the rotary jet-type mill up to 90% size grade -15 μm (powdered product) – were tested as a substitute for various grades of technical carbon and silica white when producing general rubber goods. As an illustration, in Table 4 some properties are presented in comparison with the standard properties of the experimental vulcanizates, which were produced using the silicon-carbon filler according to the mix formulation indicated in (Table 5).

In the course of the rubber compounds batching, the SC is easily



**Figure 3:** Microstructure of the Original and Pyrolyzed Rice Hulls: a – cross section of the original RH; b, c – distribution of C and Si in the original RH respectively; d – cross section of the RH pyrolyzed at 650°C; e, f – distribution of C and Si respectively in the RH pyrolyzed at 650°C.



**Figure 4:** Micrographics of Particles of the Rice Hulls Carbonizates:

a - hybrid structures that represent combination of two phases: laminar formation of the carbon phase penetrated with denser dispersed phase, probably of silicon dioxide. The supposed siliceous phase is presented by dense particles with size of 100÷200 Å, which are spread out along the translucent carbon particle.

b - aggregates formed by translucent particles of the round shape with size of 150÷200 Å.

dispersed in the natural rubber matrix. Due to this and also to the common presence, and more importantly, as stated above, to the uniform distribution of carbon- and silicon dioxide particles in the mass of each other, which were developed in the active amorphous form by pyrolysis, and due to the material's ability to be easily pulverized, application of the SC when producing the rubber technical goods, in case of partial and/or complete substitution of the technical carbon and/or silica white, and/or their mix conduces to improvement of technological properties of the rubber mixes and finished vulcanizates (increase in elasticity, improvement of the adhesive behavior and enhancement of the strength properties).

Deployment of the silicon-carbon filler in combination with

technical carbon and silica white allows producing the rubber technical articles with the specified properties. In this case the complete and even partial substitution of the standard reinforcing fillers ensures lowering the production costs of the rubber goods and tire industry articles due to relatively low (as compared with technical carbon and silica white) cost of the silicon-carbon produced from the cheap local anthropogenic raw material. Substitution of the traditional fillers with the silicon-carbon material is specified by the requirements imposed to the finished articles, and it has no provision for any changes in the production line and technological process. With a view of producing the rubbers with the required properties, it is permitted to adjust formulation of the rubber compounding.



Ingredients	Mix ingredients weight, kg				
	Standard 1	Sample 1	Sample 2	Sample 3	Sample 4
Natural rubber SKS-30 ARKM-15	0.500	0.500	0.500	0.500	0.500
Zinc white	0.040	0.040	0.040	0.040	0.040
Technical stearine	0.095	0.0095	0.095	0.095	0.095
Captax	0.010	0.010	0.010	0.010	0.010
White kaolin	0.330	0.330	0.330	0.330	0.330
Carbon black BS-100	0.160	-	-	0.160	0.160
Technical carbon P 803	0.395	0.200	0.200	0.075	0.075
SC granular	-	0.355	-	0.320	-
SC powdered	-	-	0.355	-	0.320
Plasticine	0.031	0.031	0.031	0.031	0.031
Rubrax	0.050	-	-	-	-
Sulfur	0.016	0.016	0.016	0.016	0.016
Thiuram	0.008	0.008	0.008	0.008	0.008
Total	1.635	1.585	1.585	1.585	1.585

Table 4: The Mix Formulation of the Rubber Compound for Producing the General Rubber Goods.

Properties	Vulcanizate sample				
	Standard	Sample	Sample	Sample	Sample
	1	1	2	3	4
Rupture strength, kgf/cm <sup>2</sup>	10	12	12.1	28	35.5
Ultimate elongation at rupture, %	120	78	85	100	82
Relative residual elongation after rupture, %	-	5	4	6	4
Hardness	67-80	77	76	68	77

Table 5: Physical/Mechanical Properties of General Rubber Goods.

The silicon-carbon of the rice hulls was tested with a view to use it as sorbent in various engineering processes.

It has been approved that the macroporous silicon-carbon from the rice hulls as sorbent is high active in the process of the waste-water treatment from petroleum- and oil contamination: the static volume capacity (SVC) = 50 mg/g when recovering 98% from solution that contains petroleum products and oils in the quantity of 200 mg/dm<sup>3</sup>; the dynamic volume capacity (DVC) = 80 mg/g when recovering 99.5%. In the process of the precious metals recovery from the production solutions, by its absorption characteristics on gold (SVC = 2.4 mg/g when recovering 91.4% from the solution that contained 8.1 mg/dm<sup>3</sup> of Au<sup>+</sup> ions) the activated silicon-carbon is equal to the available industrial sorbents: resin AM-2B and coking coals. When recovering rhenium from the production sulfurous solutions (pH=2), in one sorption-desorption cycle the silicon-carbon provides increase in concentration of rhenium by 2.4 times that opens prospects for its application in the sorption technology for rhenium recovery with a view to prepare the end product – ammonium perrhenate [7].

The veterinary-toxicological evaluation of the silicon-carbon from the rice hulls (we named this product “Risostim”) on the laboratory animals and by the bioindication method indicated that this product was non-toxic, and it could be used as fodder supplement and various fillers for biopreparations in veterinary. It was recommended to use “Risostim” in the ration for the laying hens and chicken broilers in concentration from 1 to 6%. In the course of the scientific-economic experiment it was specified that feeding with the mixed fodder with 3%-content of “Risostim” increased (by 8.6%) the egg-laying capacity of the laying hens and decreased the egg-breakage (by 31%), the hens mortality (by 28%) and sanitary slaughtering (by 13%) [8].

As investigations have shown, the organic condensate from the rice hulls pyrolysis has a wide range of its application. This organic condensate represents the water solution of carboxylic acids (22%),

phenols (14%), ketones (12%), cyclic aliphatic hydrocarbons (4.5%), heterocyclic compounds (4%), spirits and esters (4.5%). Owing to content in its composition of ~6% dihydroxy-benzenes (62% of which shall be pyrocatechol with its derivatives) that are used in production of high-selective ion-exchange resins, some polymeric composites with the globular structure were obtained on its basis, which had relatively high sorption capacity in relation to a number of the water contaminants (ions of As<sup>3+</sup>, Sb<sup>3+</sup> and others), and they could be used for realization of high-selective and high-speed processes of sorption and ion exchange in various branches of industry.

In addition, the pyrolyzate is a highly selective collector in relation to lead minerals in beneficiating complex rebellious lead-zinc ores [9].

The pyrolyzate is active as high-efficiency plant growth stimulant. The presowing treatment of the seeds with 0.5%-solution of pyrolyzate secures increase in productivity of the Lucerne green material by 22.4% and barley corn by 26.3% when growing them on the meadow brown-, alkali-saline-, strongly saline-, medium loamy soils against the background of the zonal agrotechnology, and the Indian corn by 28.6% when growing it on the light-chestnut-, irrigated-, weakly eroded-, medium loamy soils against the background of mineral fertilizers (N<sub>92</sub>P<sub>92</sub>).

In addition, the rice hulls pyrolyzate can be used as antiseptic; 50%-solution of it exerts the more intense inhibitory action on the tuberculosis cell culture growth *Mycobacterium bovis* St. 8 as compared with the industrial preparation “Glutar”.

The thermal decomposition of the non-condensable gases without free air delivery, owing to the presence of the saturated and unsaturated hydrocarbons in their composition (CH<sub>4</sub> ~ 40÷45 volume %; C<sub>2</sub>H<sub>4</sub> ~ 4÷6 volume %; CO<sub>2</sub> ~ 20÷22 volume %; CO ~ 22÷24 volume %; and H<sub>2</sub> ~ 4÷6 volume %), allows the additionally getting of the valuable carbon product: carbon black and high-calorific gaseous fuel.

Thus, the developed method of processing ensures the maximum possible recovery of the rice hulls with production of the silicon-carbon material and valuable by-products (organic condensate, carbon black and fuel gases), enhancement of the ecological cleanness of the process, and it opens up new potentialities for producing a number of materials of different purpose on the basis of the cheap raw material. This study has been performed under the Project of the International Science and Technology Center (the financing party – USA).

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