4th International Conference on **Electrochemistry**

June 11-12, 2018 | Rome, Italy

Electrochemical determination of caffeine content in Ethiopian coffee samples using lignin modified glassy carbon electrode

Meareg Amare and Senait Aklog Bahir Dar University, Ethiopia

ignin film was deposited at the surface of glassy carbon electrode potentiostatically. In contrast to the unmodified glassy Licarbon electrode, an oxidative peak with an improved current and overpotential for caffeine at modified electrode showed catalytic activity of the modifier towards oxidation of caffeine. Linear dependence of peak current on caffeine concentration in the range 6×10^{-6} to 100×10^{-6} molL⁻¹ with determination coefficient and method detection limit (LoD=3 s/slope) of 0.99925 and 8.37×10⁻⁷ molL⁻¹, respectively, supplemented by recovery results of 93.79–102.17% validated the developed method. An attempt was made to determine the caffeine content of aqueous coffee extracts of Ethiopian coffees grown in four coffee cultivating localities (Wonbera, Wollega, Finote Selam, and Zegie) and hence to evaluate the correlation between users preference and caffeine content. In agreement with reported works, caffeine contents (w/w%) of 0.164 in Wonbera coffee; 0.134 in Wollega coffee; 0.097 in Finote Selam coffee; and 0.089 in Zegie coffee were detected confirming the applicability of the developed method for determination of caffeine in a complex matrix environment. The result indicated that users' highest preference for Wonbera and least preference for Zegie cultivated coffees are in agreement with the caffeine content.

senaitaklog16@gmail.com